One-pot synthesis of stable phosphite ylides by three component reaction between acetylenic esters, aldehyde semicarbazones and tributyl or triethyl phosphite

Mohammad Anary-Abbasinejad*, Alireza Hassanabadi and Hossein Anaraki-Ardakani

Department of Chemistry, Islamic Azad University, Yazd Branch, P.O. Box 89195-155, Yazd, Iran

Three-component reaction between acetylenic esters, aldehyde semicarbazones and tributyl- or triethyl phosphite leads to stable crystalline phosphite ylides at one step in nearly quantitative yields.

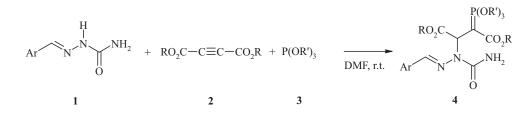
Keywords: acetylenic esters, semicarbazones, phosphite ylides, tributyl phosphite, triethyl phosphite

In recent years, there has been increasing interest in the synthesis of organophosphorus compounds. This is due to the value of such compounds in a variety of biological, industrial, and chemical synthetic uses.¹ Several methods have been described for the novel synthesis of organophosphorus compounds.^{2,3} The successful attack by nucleophilic trivalent phosphorus on a carbon atom is facilitated when the later is conjugated with a carbonyl group or when it is part of an unsaturated bond otherwise activated.1-8 There have been many studies on the reactions between trivalent phosphorus nucleophiles and unsaturated carbonyl compounds in the presence of a proton source such as an alcohol.¹ The reaction of trimethyl phosphite and dimethyl acetylenedicarboxylate (DMAD) in the presence of alcohols reported to produce phosphite ylide derivatives which are stable at low temperatures, but converted to phosphonate derivatives by warming or by treatment with water.⁹ There are other recent reports on the reaction between phosphites and acetylenic esters in the presence of an acidic organic compound, all of them proceeding through a phosphite ylide intermediate.¹⁰⁻¹² However, this intermediate has not been isolated nor characterised in any of these works and usually hydrolysed or rearranged to the corresponding phosphonates. In another work, we have reported that the three-component reaction between trimethyl phosphite, acetylenic esters and aldehyde semicarbazones, leads to phosphonate derivatives passing from phosphite ylide intermediate.¹³ In order to explore the scope of this reaction, we decided to investigate the same

reaction with other phosphites such as tributyl phosphite, triethyl phosphite or triphenyl phosphite. Thus, the reaction between tributyl or triethyl phosphite **3** and acetylenic ester **2** in the presence of aldehyde semicarbazone **1** was carried out in DMF at room temperature. The only isolated product was stable crystalline phosphite ylide **4** obtained in excellent yield (Scheme 1).

The three-component reaction between triethyl phosphite, benzaldehyde semicarbazone and diethyl acetylendicarboxylate also afforded the phosphite ylide **4f**. However, no product was isolated from the similar reaction between triphenyl phosphite, acetylenic esters and semicarbazones except the starting semicarbazone.

The structures of compounds **4a–f** result from their IR, ¹H, ¹³C, and ³¹P NMR spectra. The mass spectra of the ylides **4** are fairly similar and display molecular ion peaks. The ¹H NMR spectrum of **4a** shows the presence of three butoxy groups. It also exhibits two sharp lines at $\delta = 3.44$ and 3.60 ppm for the protons of two methyls of methoxy groups and a doublet (${}^{3}J_{\rm HP} = 6 \,{\rm H_{Z}}$) at 6.36 for methine proton which is coupled with phosphorus atom. The HC=N proton appears at $\delta = 7.28$ –7.65 ppm. A broad singlet is observed at $\delta = 6.75$ ppm for NH₂ protons which was disappeared after the addition of D₂O to the d₆-DMSO solution of **4a**. The ³¹P NMR spectrum of compound **4a** displays a signal at 53.6 ppm. This shift is similar to those observed for other stable phosphite ylides.⁹



4	Ar	R	R'	%Yield*
a	Phenyl	Me	Bu	95
b	$p\text{-}\mathrm{Cl}\text{-}\mathrm{C}_{6}\mathrm{H}_{4}$	Et	Bu	97
c	<i>m</i> -CH ₃ O-C ₆ H ₄	Me	Bu	92
d	m-CH ₃ O-C ₆ H ₄	Et	Bu	88
e	$\begin{array}{c} m\text{-}\mathrm{CH}_{3}\mathrm{O}\text{-}\mathrm{C}_{6}\mathrm{H}_{4}\\ m\text{-}\mathrm{CH}_{3}\mathrm{O}\text{-}\mathrm{C}_{6}\mathrm{H}_{4}\\ m\text{-}\mathrm{CH}_{3}\mathrm{O}\text{-}\mathrm{C}_{6}\mathrm{H}_{4} \end{array}$	t-Bu	Bu	94
f	Phenyl	Et	Et	95
* Isolated yield.				

Scheme 1

* Correspondent. E-mail: mohammadanary@yahoo.com

On the basis of the well-established chemistry of trivalent phosphorus nucleophiles¹⁻⁷ it is reasonable to assume that the ylide **4** results from the initial addition of trialkyl phosphite to the acetylenic ester and subsequent protonation of the 1:1 adduct by the NH-acid (Scheme 2). Then, the positively charged ion **6** is attacked by the anion of the NH-acid **5** to form the phosphite ylide **4**.

In summary, stable crystalline phosphite ylides may be prepared by a simple, one-pot three-component reaction between acetylenic esters, aldehyde semicarbazones, and tributyl phospite or triethyl phosphite. The present method carries the advantage that not only is the reaction performed under neutral conditions but also that the substances can be mixed without any activation or modification.

Experimental

Melting points were determined with an electrothermal 9100 apparatus. Elemental analyses were performed using a Heraeus CHN-O-Rapid analyser. Mass spectra were recorded on a FINNIGAN-MAT 8430 mass spectrometer operating at an ionisation potential of 70 eV. IR spectra were recorded on a Shimadzu IR-470 spectrometer. ¹H, ¹³C, and ³¹P NMR spectra were recorded on Bruker DRX-500 Avance spectrometer at solution in d₆-DMSO using TMS as internal standard or 85% H₃PO₄ as external standard. The chemicals used in this work purchased from fluka (Buchs, Switzerland) and were used without further purification.

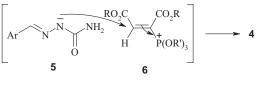
General procedure

To a magnetically stirred solution of phosphite (2 mmol) and semicarbazone (2 mmol) in DMF (10 ml) was added drop-wise a mixture of dialkyl acetylenedicarboxylate (2 mmol) in DMF (3 ml) at room temperature over 2 min. The reaction mixture was then stirred for 24 h. Water (50 ml) was added and the mixture was extracted by dichloromethane (3×20 ml). The organic phase was washed with water (3×20 ml) and dried over anhydrous sodium sulfate. Solvent was evaporated and the residue was crystallised from ethyl acetate-hexane mixture.

Dimethyl 2-(tributoxyphosphoranylidene)-3-[5-phenyl-2-oxol,3,4-triazapent-4-en-3-yl]succinat (4a): Colourless crystals, m.p. 77–80°C, IR (KBr) (v_{max} cm⁻¹): 3480, 3345 (NH₂), 1746, 1690 (2 C=0, ester), 1645 (C=0, amid). Analyses: Calcd. for C₂₆H₄₂N₃O₈P: C, 56.21; H, 7.62; N, 7.56%. Found: C, 55.9; H, 7.6; N, 7.8. MS (m/z,%): 555 (6). ¹H NMR (500 MHz, d₆-DMSO): δ 0.77 (9 H, t, 3 CH₃), 1.19 (6 H, sextet, 3 CH₂), 1.49 (6 H, quintet, 3 CH₂), 3.44 (3 H, s, OCH₃), 3.60 (3 H, s, OCH₃), 3.88 (6 H, t, POCH₂), 6.36 (1 H, d, ³J_{HP} = 16 Hz, CHN), 6.65 (2 H, broad s, NH₂), 7.28–7.65 (5 H, m, 5 CH aromatic), 7.93 (1 H, s, N=CH). ¹³C NMR (125.8 MHz, d₆-DMSO): δ 13.70 (3 CH₃), 18.41(3 CH₂), 31.76 (d, ³J_{cp} = 7 Hz, 3 CH₂), 41.46 (d, ¹J_{cp} = 144 Hz, C = P), 49.85 and 52.23 (2 OCH₃), 54.87(d,²J_{cp} = 14 Hz, CHN), 67.89 (d, ²J_{cp} = 6 Hz, 3 OCH₂), 126.77, 128.91, 136.15 and 143.56 (aromatic), 157.29 (C=N), 159.32 (C=O), 169.05 (d, ²J_{cp} = 19 Hz, C=O ester), 171.78 (d, ³J_{cp} = 19 Hz, C=O ester). ³¹P NMR (202.5 MHz, d₆-DMSO): δ 53.6.

Diethyl 2-(tributoxyphosphoranylidene)-3-[5-(4-chlorophenyl)-2-oxo-1,3,4-triazapent-4-en-3-yl]succinat (4b): Colourless crystals, m.p. 117–119°C. IR (KBr) (v_{max}, cm⁻¹): 3495, 3360 (NH₂), 1746, 1688 (2 C=O, ester), 1642 (C=O, amid). Analyses: Calcd. for C₂₈H₄₅ClN₃O₈P: C, 54.41; H, 7.34; N, 6.80%. Found: C, 54.3; H, 7.3; N, 6.9 MS (*m*/*z*,%): 618 (5). ¹H NMR (500 MHz, d₆-DMSO): δ 0.78 (9 H, t, 3 CH₃), 1.07 and 1.13 (6 H, 2 t, 2 CH₃), 1.18 (6 H, sextet, 3 CH₂), 1.49 (6 H, quintet, 3 CH₂), 3.88 (6 H, t, POCH₂), 3.94 and 4.04 (4-H, m, 2 OCH₂), 6.34 (1 H, d, ³J_{HP} = 16 Hz, CHN), 6.63 (2 H, broad s, NH₂), 7.40–7.68 (4 H, m, 4 CH aromatic), 7.96 (1 H, s, N=CH). ¹³C NMR (125.8 MHz, d₆-DMSO): δ 13.71 (3 CH₃), 14.43 and 15.35 (2 CH₃ of ethy groups), 18.44 (s, 3 CH₂), 31.81 (d, ³J_{cp} = 7 Hz, 3 CH₂), 41.46 (d, ¹J_{cp} = 143 Hz, C = P), 54.97 (d, ²J_{cp} = 14 Hz, CHN), 58.02 and 60.66 (2 OCH₂), 67.74 (d, ²J_{cp} = 6 Hz, 3 OCH₂), 128.38, 128.97, 135.16 and 142.12 (aromatic), 157.24 (C=N), 162.71 (C=O), 168.69 (d, ²J_{cp} = 19 Hz, C=O ester), 171.16 (d, ³J_{cp} = 19 Hz, C=O ester). ³¹P NMR (202.5 MHz, d₆-DMSO): δ 54.3

Dimethyl 2-(tributoxyphosphoranylidene)-3-[5-(3-methoxyphenyl-2-oxo-1,3,4-triazapent-4-en-3-yl]succinat (4c): Colourless crystals, m.p. 107–110°C. IR (KBr) (v_{max}, cm⁻¹): 3475, 3340 (NH₂), 1743, 1685 (2 C=O, ester), 1640 (C=O, amid). Analyses: Calcd. for C₂₇H₄₄N₃O₉P: C, 55.37; H, 7.57; N, 7.18%. Found: C, 55.2; H, 7.5; N, 7.4. MS (*m*/*z*,%): 585(7). ¹H NMR (500 MHz, d₆-DMSO): δ 0.78



Scheme 2

(9 H, t, 3 CH₃), 1.19 (6 H, sextet, 3 CH₂), 1.49 (6 H, quintet, 3 CH₂), 3.44 (3 H, s, OCH₃), 3.59 (3 H, s, OCH₃), 3.77 (3 H, s, OCH₃), 3.88 (6 H, t, POCH₂), 6.35 (1 H, d, ${}^{3}J_{HP}$ = 16 Hz, CHN), 6.67 (2 H, broad s, NH₂), 7.12–7.34 (4 H, m, 4 CH aromatic), 7.89 (1H, s, N=CH). ${}^{13}C$ NMR (125.8 MHz, d₆-DMSO): δ 13.75 (s, 3 CH₃), 18.51 (s, 3 CH₂), 31.82 (d, ${}^{3}J_{cp}$ = 7 Hz, 3 CH₂), 41.51 (d, ${}^{1}J_{cp}$ = 145 Hz, C = P), 49.84 and 52.24 (2 OCH₃), 55.25 (d, ${}^{2}J_{cp}$ = 14 Hz, CHN), 55.59 (s, OCH₃), 67.90 (d, ${}^{2}J_{cp}$ = 6 Hz, 3 OCH₂), 110.90, 115.57, 119.94, 129.97, 138.27 and 143.69 (aromatic), 157.29 (C=N), 159.98 (C=O), 169.12 (d, ${}^{2}J_{cp}$ = 19 Hz, C=O ester), 171.81 (d, ${}^{3}J_{cp}$ = 19 Hz, C=O ester). ${}^{31}P$ NMR (202.5 MHz, d₆-DMSO): δ 54.0.

Diethyl 2-(tributoxyphosphoranylidene)-3-[5-(3-methoxyphenyl)-2-oxo-1,3,4-triazapent-4-en-3-yl]succinat (4d): Colourless crystals, m.p. 72–75°C. IR (KBr)(v_{max} , cm⁻¹): 3540, 3355(NH₂), 1742, 1698 (2 C=0, ester), 1639 (C=0, amid). Analyses: Calcd. for C₂₉H₄₈N₃O₉P: C, 56.76; H, 7.88; N, 6.85%. Found: C, 56.6; H, 7.8; N, 7.0. MS (*m*/*z*,%): 613(9). ¹H NMR (500 MHz, d₆-DMSO): δ 0.79 (9 Ht, 3 CH₃), 1.07 and 1.14 (6H, 2t, 2 CH₃ of ethyl groups), 1.19 (6H, sextet, 3 CH₂), 1.50 (6 H, quintet, 3 CH₂), 3.75 (3 H, s, OCH₃), 3.89 (6 H, t, POCH₂), 3.95 and 4.07 (4 H, m, 2 OCH₂), 6.33 (1 H, d, ³J_{HP} = 16 Hz, CHN), 6.66 (2 H, broad s, NH₂), 7.13–7.64 (4 H, m, 4 CH aromatic), 7.95 (1H, s, N=CH). ¹³C NMR (125.8 MHz, d₆-DMSO): δ 13.71 (s, 3 CH₂), 14.43 and 15.36 (2 CH₃), 18.53 (s, 3 CH₂), 31.85 (d, ³J_{cp} = 7 Hz, 3 CH₂), 41.42 (d, ¹J_{cp} = 141 Hz, C = P), 54.95 (d, ²J_{cp} = 14 Hz, CHN), 55.49 (OCH₃), 57.96 and 60.72 (2 OCH₂), 67.73 (d, ²J_{cp} = 6 Hz, 3 OCH₂), 110.83, 115.38, 119.97, 129.91, 139.52 and 143.70 (aromatic), 157.28 (C=N), 159.97 (C=O), 168.65 (d, ²J_{cp} = 19 Hz, C=O ester), 171.28 (d, ³J_{cp} = 19 Hz, C=O ester). ³¹P NMR (202.5 MHz, d₆-DMSO): δ 53.7.

Di-t-butyl 2-(*tributoxyphosphoranylidene*)-3-[5-(3-*methoxyphenyl*-2-*oxo*-1,3,4-*triazapent*-4-*en*-3-*yl*]*succinate* (4e): Colourless crystals, m.p. 103–106°C, IR(KBr) (v_{max} , cm⁻¹): 3400, 3290 (NH₂), 1737, 1689 (2 C=O, ester), 1641 (C=O, amid). Analyses: Calcd. for C₃₃H₅₆N₃O₉P: C, 59.18; H, 8.43; N, 6.27%. Found: C, 58.9; H, 8.4; N, 6.1%. MS (*m/z*,%): 669 (10). ¹H NMR (500 MHz, d₆-DMSO): δ 0.79 (9 H, t, 3 CH₃), 1.19 (6 H, sextet, 3 CH₂), 1.37 (s, 9 H), 1.39 (s, 9 H), 1.50 (6 H, quintet, 3 CH₂), 3.75 (3 H, s, OCH₃), 3.86 (6 H, t, POCH₂), 6.18 (1H, d, ³J_{HP} = 16 Hz, CHN), 6.72 (2 H, broad s, NH₂), 7.18–7.34 (4 H, m, 4 CH aromatic), 8.05 (1 H, s, N=CH). ¹³C NMR (125.8 MHz, d₆-DMSO): δ 13.82 (3 CH₃), 18.52 (3 CH₂), 28.24 and 29.06 (6 CH₃ of 2 t-Bu), 31.97 (d, ³J_{cp} = 7 Hz, 3 CH₂), 41.41 (d, ¹J_{cp} = 141 Hz, C = P), 55.43 (d, ²J_{cp} = 14 Hz, CHN), 55.62 (OCH₃), 67.18 (d, ²J_{cp} = 6 Hz, 3 OCH₂), 76.98 and 79.63 (2 O–C(CH₃)₃), 110.97, 115.31, 119.80, 129.80, 137.80 and 143.44 (aromatic), 157.31 (C=N), 159.91 (C=O), 168.25 (d, ²J_{cp} = 19 Hz, C=O ester), 169.99 (d, ³J_{cp} = 19 Hz, C=O ester). ³¹P NMR (202.5 MHz, d₆-DMSO): δ 54.5.

Diethyl 2-(triethoxyphosphoranylidene)-3-[5-phenyl-2-oxo-1,3,4-triazapent-4-en-3-yl]succinat (**4f**): Colourless crystals, m.p. 78–81°C, IR (KBr) (v_{max} , cm⁻¹): 3540,3355 (NH₂), 1742, 1698 (2 C=O, ester), 1639 (C=O, amid). Analyses: Calcd. for C₂₂H₃₄N₃O₈P: C, 52.90; H, 6.86; N, 8.41%. Found: C, 52.9; H, 6.7; N, 8.6%. MS (m/z,%): 499 (11). ¹H NMR (500 MHz, d₆-DMSO): δ 0.98 1.12 (15 H, m, 5 CH₃), 3.95 – 4.05 (10 H, m, 5 OCH₂), 6.28 (1 H, d, ³J_{HP} = 16 Hz, CHN), 6.54 (2 H, broad s, NH₂), 7.32 – 7.66 (5 H, m, 5 CH aromatic), 7.97 (1H, s, N=CH). ¹³C NMR(125.8 MHz, d₆-DMSO): δ 13.75 (d, ³J_{CP} = 6 Hz, 3 CH₃), 14.21 and 14.33 (2 CH₃), 41.46 (d, ¹J_{cp} = 141 Hz, C=P), 54.73 (d,²J_{cp} = 14 Hz, CHN), 62.89 and 93.22 (2 OCH₂), 63.23 (d, ²J_{cp} = 7 Hz, 3 POCH₂), 126.42, 128.90, 137.11 and 143.56 (aromatic), 157.29 (C=N), 159.32 (NC=O), 169.05 (d, ²J_{cp} = 19 Hz, C=O ester), 171.78 (d, ³J_{cp} = 19 Hz, C=O ester). ³¹P NMR(202.5 MHz, d₆-DMSO): δ 54.2.

Received 29 June 2007; accepted 25 July 2007 Paper 07/4720 doi: 10.3184/030823407X236372

References

- 1 D.E.C. Corbridge, *Phosphorus an Outline of the Chemistry, Biochemistry, and Uses*, 5th edn. Elsevier, Amsterdam, 1995.
- 2 R. Engel, *Synthesis of Carbon-Phosphorus Bonds*, CRC Press, Boca Raton, FL, 1988.

- 3 J.I.G. Cadogan, Organophosphorus in Organic Synthesis, Academic Press, New York, 1979.
- 4 O.I. Kolodiazhnyi, Russ. Chem. Rev., 1997, 66, 225.
- 5 H.J. Bestmann and R. Zimmermann, Top. Curr. Chem., 1983, 109, 85.
- 6 B.E. Maryano and A.B. Reits, Chem. Rev., 1989, 89, 863.
- 7 J.C. Tebbyin, *Phosphorus-31 NMR Spectroscopy in Stereochemical Analysis*, J.C. Verkede and L.D. Quin (eds), VCH, Weinheim, chap. 1, pp 1-60, 1987
- 8 M.V. George, S.K. Khetan and R.K. Gupta, *Adv. Heterocycl. Chem.*, 1976, 19, 354.
- 9 A.W. Johnson, W. C. Kaska, A. O. Starzewski and D. A. Dixon, *Ylides and Imines of Phosphorus*, John Wiley & Sons, New York, 1993, pp. 386-387.
- 10 I. Yavari and M. Anary-Abbasinejad, Org. Biomol. Chem., 2003, 3, 560.
- 11 M.T. Maghsoodlou, S.M.H. Khorassani, R. Heydari and F.R. Charati, J. Chem. Research, 2006, 364.
- 12 M.T. Maghsoodlou, S.M.H. Khorassani, M.K. Rofouei, S.R. Adhamdoust and M. Nassiri, Arkivoc, 2006, 143.
- 13 M. Anary-abbasinejad and N. Ascarrian, J. Chem. Res., 2007, 11.